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Pre-lab Assignment/Questions *Note- this pre-lab must be finished before you come to lab.
1. Calculate the molar mass of citric acid $C_6H_8O_7$.
2. Calculate the volume of 2.5 M NaOH needed to make 200 mL of 0.04 M NaOH.
3. The balanced equation for KHP and NaOH is given in the introduction to this lab.

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What is distinctive about the ways specific disciplines are traditionally taught, and what kinds of learning do they promote? Do they inspire the habits of the discipline itself, or do they inadvertently contradict or ignore those disciplines? By analyzing assumptions about often unexamined teaching practices, their history, and relevance in contemporary learning contexts, this book offers teachers a fresh way to both think about their impact on students and explore more effective ways to engage students in authentic habits and practices. This companion volume to *Exploring Signature Pedagogies* covers disciplines not addressed in the earlier volume and further expands the scope of inquiry by interrogating the teaching methods in interdisciplinary fields and a number of professions, critically returning to Lee S. Shulman's origins of the concept of signature pedagogies. This volume also differs from the first by including authors from across the United States, as well as Ireland and Australia. The first section examines the signature pedagogies in the humanities and fine arts fields of philosophy, foreign language instruction, communication, art and design, and arts entrepreneurship. The second section describes signature pedagogies in the social and natural sciences: political science, economics, and chemistry. Section three highlights the interdisciplinary fields of

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Ignatian pedagogy, women's studies, and disability studies; and the book concludes with four chapters on professional pedagogies - nursing, occupational therapy, social work, and teacher education - that illustrate how these pedagogies change as the social context changes, as their knowledge base expands, or as online delivery of instruction increases.

Retaining the concise, to-the-point presentation that has already helped thousands of students move beyond memorization to a true understanding of the beauty and logic of organic chemistry, this Seventh Edition of John McMurry's *FUNDAMENTALS OF ORGANIC CHEMISTRY* brings in new, focused content that shows students how organic chemistry applies to their everyday lives. In addition, redrawn chemical structures and artwork help students visualize important chemical concepts, a greater emphasis on biologically-related chemistry (including new problems) helps them grasp the enormous importance of organic chemistry in understanding the reactions that occur in living organisms, and new End of Chapter problems keyed to OWL allow them to work text-specific problems online. Lastly, , for this edition, John McMurry reevaluated and revised his writing at the sentence level to ensure that the book's explanations, applications, and examples are more student-friendly, relevant, and motivating than ever before. Important Notice: Media content

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Steve and Susan Zumdahl's texts focus on helping students build critical thinking skills through the process of becoming independent problem-solvers. They help students learn to think like a chemists so they can apply the problem solving process to all aspects of their lives. In CHEMISTRY: AN ATOMS FIRST APPROACH, the Zumdahls use a meaningful approach that begins with the atom and proceeds through the concept of molecules, structure, and bonding, to more complex materials and their properties. Because this approach differs from what most students have experienced in high school courses, it encourages them to focus on conceptual learning early in the course, rather than relying on memorization and a plug and chug method of problem solving that even the best students can fall back on when confronted with familiar material. The atoms first organization provides an opportunity for students to use the tools of critical thinkers: to ask questions, to apply rules and models and to evaluate outcomes. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

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presentation of lab techniques available for organic chemistry students - and the least expensive. It combines specific instructions for 3 different kinds kinds of laboratory glassware and offers extensive coverage of spectroscopic techniques and a strong emphasis on safety issues.

Updated and price drop Fall 2020

For students, DIY hobbyists, and science buffs, who can no longer get real chemistry sets, this one-of-a-kind guide explains how to set up and use a home chemistry lab, with step-by-step instructions for conducting experiments in basic chemistry -- not just to make pretty colors and stinky smells, but to learn how to do real lab work: Purify alcohol by distillation Produce hydrogen and oxygen gas by electrolysis Smelt metallic copper from copper ore you make yourself Analyze the makeup of seawater, bone, and other common substances Synthesize oil of wintergreen from aspirin and rayon fiber from paper Perform forensics tests for fingerprints, blood, drugs, and poisons and much more From the 1930s through the 1970s, chemistry sets were among the most popular Christmas gifts, selling in the millions. But two decades ago, real chemistry sets began to disappear as manufacturers and retailers became concerned about liability. ,em>The Illustrated Guide to Home Chemistry Experiments steps up to the plate with lessons on how to equip your home

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chemistry lab, master laboratory skills, and work safely in your lab. The bulk of this book consists of 17 hands-on chapters that include multiple laboratory sessions on the following topics: Separating Mixtures Solubility and Solutions Colligative Properties of Solutions Introduction to Chemical Reactions & Stoichiometry Reduction-Oxidation (Redox) Reactions Acid-Base Chemistry Chemical Kinetics Chemical Equilibrium and Le Chatelier's Principle Gas Chemistry Thermochemistry and Calorimetry Electrochemistry Photochemistry Colloids and Suspensions Qualitative Analysis Quantitative Analysis Synthesis of Useful Compounds Forensic Chemistry With plenty of full-color illustrations and photos, Illustrated Guide to Home Chemistry Experiments offers introductory level sessions suitable for a middle school or first-year high school chemistry laboratory course, and more advanced sessions suitable for students who intend to take the College Board Advanced Placement (AP) Chemistry exam. A student who completes all of the laboratories in this book will have done the equivalent of two full years of high school chemistry lab work or a first-year college general chemistry laboratory course. This hands-on introduction to real chemistry -- using real equipment, real chemicals, and real quantitative experiments -- is ideal for the many thousands of young people and adults who want to experience the magic of chemistry.

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In their professional dreams, chemistry teachers imagine eager and self-sufficient students whose curiosity motivates their scientific explorations. Joan Gallagher-Bolos and Dennis Smithenry have realized this vision in their chemistry classrooms, and in *Teaching Inquiry-Based Chemistry*, they demonstrate how you can make student-led inquiry happen in yours. *Teaching Inquiry-Based Chemistry* retraces an entire year's curriculum to show you how the authors weave constructivist theory into every lesson without sacrificing content. You will discover how slowly increasing the complexity of projects while gradually shifting the responsibility for learning to class members builds success upon success until students are ready to formulate and execute a three-week, end-of-year project where they function as a fully independent scientific community. Plus *Teaching Inquiry-Based Chemistry* is loaded with features that help you implement student-centered teaching immediately, including: proven instructional strategies examples of successful units from the authors' own curricula graphic organizers that guide you through creating an inquiry-driven classroom discussions of meeting NSES's inquiry standards through inquiry-based teaching in-depth examples of student journals and projects Get ready to make your ideal classroom a reality and find a fresh way of teaching the chemistry you know so

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well. Read Teaching Inquiry-Based Chemistry and discover how helping your students capitalize on their innate scientific curiosity will lead you to new levels of professional and personal satisfaction.

Class-tested by thousands of students and using simple equipment and green chemistry ideas, UNDERSTANDING THE PRINCIPLES OF ORGANIC CHEMISTRY: A LABORATORY COURSE includes 36 experiments that introduce traditional, as well as recently developed synthetic methods. Offering up-to-date and novel experiments not found in other lab manuals, this innovative book focuses on safety, gives students practice in the basic techniques used in the organic lab, and includes microscale experiments, many drawn from the recent literature. An Online Instructor's Manual available on the book's instructor's companion website includes helpful information, including instructors' notes, pre-lab meeting notes, experiment completion times, answers to end-of-experiment questions, video clips of techniques, and more. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

Eschewing the usual mathematical explanations for physics phenomena, this approachable reference explains complicated scientific concepts in plain English that everyone can

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understand. Tackling the big issues such as gravity, magnetism, sound, and what really happens in the Large Hadron Collider, this engaging look at physics also spells out why cats always land on their feet, why people appear to have red eyes in photographs, and the real danger of looking at an eclipse. For everyone who ever wondered how a light bulb works or how squirrels avoid electrocution on the power lines, this handbook supplies answers on the physics of everyday life and examines the developments in the exploration of subatomic particles. In addition to the question-and-answer section, an addendum of facts about physicists explains what the Nobel prize is and who has won it, and tells the story of the scientist who was incarcerated for agreeing with Copernicus. Answers more than eight hundred questions about physics, ranging from everyday life applications to the latest explorations in the field.

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